PREPARATION AND REACTIONS OF TRIPHENYLARSORANYLIDENEKETENE

H.J. Bestmann and R.K. Bansal

Institut für Organische Chemie, Universität Erlangen-Nürnberg, Henkestr. 42, D 852 Erlangen, Germany.

Abstract: Preparation and some reactions of the first cumulated ylide of arsenic, viz. triphenylarsoranylideneketene are described.

The cumulated-phosphonium ylides (<u>1</u>) form a unique group within the ylides, which have been recently used as very versatile synthones in organic synthesis.¹ The first cumulated-phosphonium ylide (<u>la</u>) was prepared by Matthews and Birum² from the thermolysis of the adduct obtained from the interaction of CO₂ with hexaphenylcarbodiphosphorane. A simple method was

(C ₆ H ₅) ₃ P=C=C≠X	(C ₆ H ₅) ₃ P=CH-C-XCH ₃
<u>1</u>	<u>2</u>
a, X=0 b, X=S c, X=N-Ar	X=0;S

later developed for the preparation of this ylide and its thioanalogue, which involved β -elimination of CH₃XH (X=0,S) from the ylide, <u>2</u> by sodium bis(trimethylsilyl)amide.³ We have now extended this method to the synthesis of the first cumulated ylide of arsenic, viz. triphenylarsoranylideneketene (3).

All the reactions were carried out under nitrogen. Methoxycarbonylmethylenetriphenylarsorane⁴ (<u>4</u>, m.p. 161-63[°]; 1 equiv.) was stirred in toluene with sodium bis(trimethylsilyl)amide (<u>5</u>,1.1 equiv.) overnight at room temperature. The reaction mixture was filtered and the filtrate was concentrated under reduced pressure to a syrupy mass which on crystallization from toluene/ether afforded pale yellow crystals of <u>3</u>, m.p. 141-43[°] (yield, 60%). IR (KBr), strong absorption at 2030 cm⁻¹ (\mathcal{V} C=C=O); ¹H-NMR (CDCl₃), S 7.3-8.0 ppm (aromatic H); ¹³C-NMR, (CDCl₃), -5.5 ppm (C_{ac}), 128.5-133.5 ppm (aromatic C), 150.0 ppm (C_B=O); M[±] = 346 m/e. Scheme I

$$(C_{6}H_{5})_{3}As=CH-COCH_{3} \longrightarrow NaN[Si(CH_{3})_{3}]_{2} \longrightarrow \frac{4}{5}$$

$$(C_{6}H_{5})_{3}As=C=C=0 + NaOCH_{3} + HN[Si(CH_{3})_{3}]_{2}$$

$$3$$

In the light of various interesting reactions shown by the cumulated-phosphonium ylides¹, a few representative reactions of the ylide, $\frac{3}{2}$ have been carried out.

To the stirred solution of $\underline{3}$ (2 equiv.) in benzene was added methyl iodide (1 equiv.) dropwise and the resulting solution was stirred for 72 hrs. at room temperature. A pale yellow solid separated out which was filtered, washed with ether and dried, m.p. $163-65^{\circ}$ (decomp.) (yield, 50%). IR (KBr), 1760 cm⁻¹ (w) and 1640 cm⁻¹ (s); ¹H-NMR (CDCl₃), $\underline{5}^{2.0}$ ppm (3H, CH₃), 7.3-8.0 ppm (30 H, aromatic H); ¹³C-NMR (CDCl₃), $\underline{5}^{16.7}$ ppm (CH₃), 72.9 ppm (C₁), 82.7 ppm (C₃), 120.0 - 134.5 (aromatic C) and 181.7 ppm (C₂ and C₄); Mass, 707 m/e (M[±]-J, i.e. C₄₁H₃₃0₂As₂). On the basis of these results, structure <u>6</u> was assigned to this compound which is formed as shown in Scheme II. Thus, the reaction of <u>3</u> with methyl iodide proceeds similar to that of the corresponding phosphonium ylide, <u>1a</u>.

Scheme II

$$\underline{3} + CH_{3}I \longrightarrow [(C_{6}H_{5})_{3}As - C = C = 0] I^{\bullet}$$

$$(C_{6}H_{5})_{3}As = C = C = 0$$

$$(C_{6}H_{5})_{3}As = C = C = 0$$

$$(C_{6}H_{5})_{3}As = \frac{CH_{3}}{2} = \frac{CH_{3}}$$

6

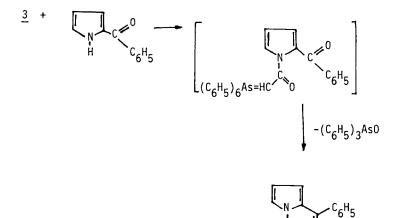
To the ylide, <u>3</u> (1 equiv.) suspended in toluene was added ethanethiol (1 equiv.) and the solution was heated at 70-80° for 4 hrs. The solvent was then removed and the residue on crystallization from ethyl acetate/ether afforded pale yellow crystals of thioethyl-carboxymethylenetriphenylarsorane (<u>7</u>), m.p. 168-170° (yield, 74%); IR (KBr, \mathcal{P}_{CO} 1550 cm⁻¹; ¹H-NMR(CDCl₃), **5**1.3 ppm (t, 3H), 2.75 ppm (q, 2H), 3.93 ppm (s, 1H), 7.3 - 7.9 ppm (m, 15H). Scheme III

$$\underline{3} + \mathrm{CH}_3\mathrm{CH}_2\mathrm{SH} \longrightarrow (\mathrm{C}_6\mathrm{H}_5)_3\mathrm{As}=\mathrm{CHCOSCH}_2\mathrm{CH}_3$$

Dale and Froeyen⁴ have reported the temperature-dependence of PMR spectra of methoxycarbonylmethylenetriphenylarsoranes. A similar phenomenon has been observed in the case of analogous phosphonium ylides.⁶ In the view of this, PMR spectrum of the ylide, <u>7</u> was studied (in pyridine) at different temperatures. On increasing the temperature, the signal for <u>CH</u> at 3.93 ppm broadened slowly and at 110⁰ a clear tendency was observed for splitting each peak into two.

To the ylide 3, (1 equiv.) dissolved in methylene chloride was added 2-benzoylpyrrole (1 equiv.) in methylene chloride slowly at room temp. A deep red colour developed within 1 hour. The solution was, however, stirred overnight. The solvent was removed under reduced pressure and the semi-solid on chromatography over silica-gel (eluent, benzene) afforded a red compound (8) which was recrystallized from toluene/pet. ether, m.p. 78-80° (yield, 85%). IR (KBr), $\boldsymbol{\gamma}_{\rm CO}$ 1730 cm⁻¹. It was found to be identical with 1-phenylpyrrolizin-3-one⁷.

Scheme IV



8

3841

3842

The ylide, 3 like the analogous phosphonium ylides is expected to show other interesting reactions which are being investigated.

The financial support from Alexander von Humboldt Foundation is gratefully acknowledged.

References

- Review, H.J. Bestmann, Angew. Chem. <u>89</u>, 367 (1977); Angew. Chem. Int. Ed. Engl. <u>16</u>, 349 (1977) and the references cited therein.
- 2. C.N. Matthews and G.H. Birum, Tetrahedron Lett. 5707 (1966).
- 3. H.J. Bestmann and D. Sandmeier, Chem. Ber. 113, 274 (1980).
- 4. A.J. Dale and P. Froeyen, Acta Chem. Scand. 25, 1452 (1971).
- 5. G.H. Birum and C.N. Matthews, J. Am. Chem. Soc. <u>90</u>, 3842 (1968).
- 6.a. H.J. Bestmann, G. Joachim, I. Lengyel, J.F.M. Oth, R. Merenyi and H. Weitkamp, Tetrahedron Lett. 3355 (1966); b.F.J. Randall and A.W. Johnson, Tetrahedron Lett. 2841 (1968); c.H.I. Zeliger, J.P. Snyder and H.J. Bestmann, Tetrahedron Lett. 2199 (1969); d.D.M. Crouse, A.T. Wehman and E.E. Schweizer, Chem. Commun. 866 (1968).
- H.J. Bestmann, G. Schmid and D. Sandmeier, Angew. Chem. <u>88</u>, 92 (1976); Angew. Chem. Int. Ed. Engl. 15, 115 (1976).

(Received in Germany 2 July 1981)